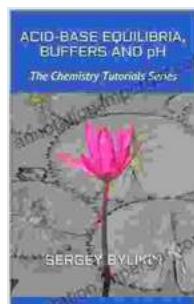


Master Acid-Base Equilibria, Buffers, and pH with the Comprehensive Guide: Chemistry Tutorials Series

Understanding acid-base equilibria, buffers, and pH is crucial for various scientific disciplines and real-world applications. This comprehensive book in the Chemistry Tutorials Series delves deep into these fundamental concepts, providing a thorough theoretical foundation and practical insights.

Chapter Outline

Chapter 1: Acids, Bases, and pH- Definitions and properties of acids and bases - The pH scale and its significance



Acid-Base Equilibria, Buffers and pH (The Chemistry Tutorials Series) by Sergey Bylikin

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Chapter 2: Acid-Base Equilibria- Proton transfer reactions - Equilibrium constants and Ka values - Acid dissociation and conjugate base formation

Chapter 3: Buffers- to buffers - The buffer capacity and Henderson-Hasselbalch equation - Applications of buffers in biological and chemical systems

Chapter 4: pH Measurement- Principles of pH measurement - Electrodes and pH meters - Calibrating and using pH meters accurately

Chapter 5: Acid-Base Titrations- Equivalence point and titration curves - Strong and weak acid-base titrations - Applications in quantitative analysis

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Comprehensive Coverage:- In-depth exploration of acid-base equilibria, buffers, and pH concepts - Covers both theoretical and applied aspects

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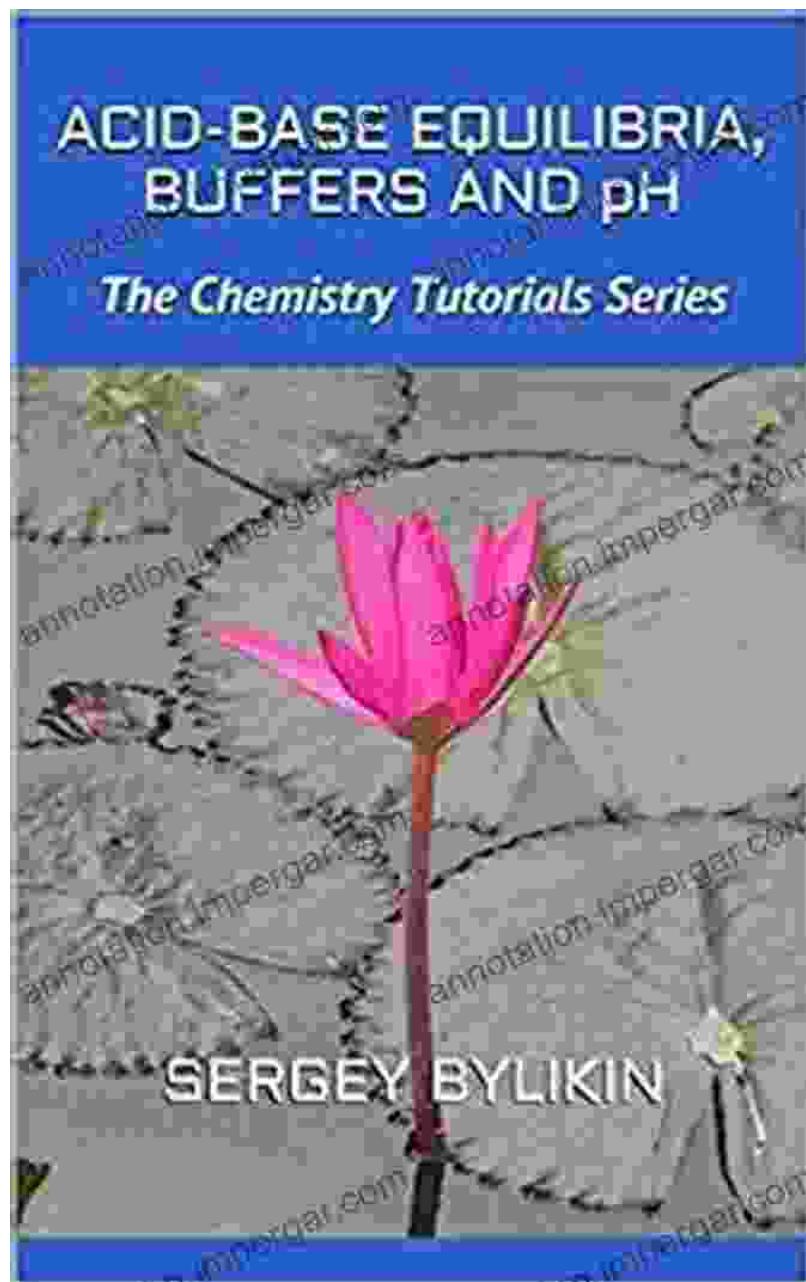
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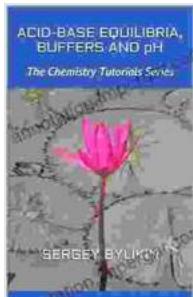
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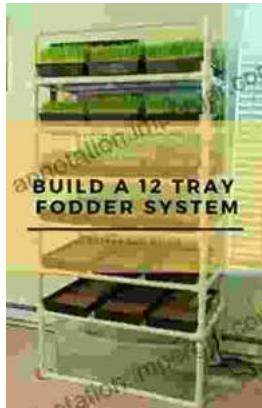
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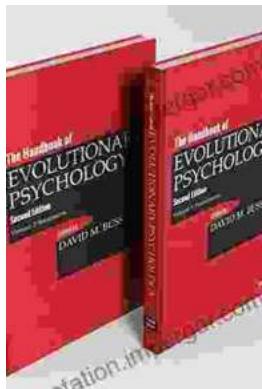


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